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Chiral thiophosphoramide and selenophosphoramide ligands in the Cu(I)-promoted catalytic enantioselective 1,3-dipolar cycloaddition reactions of azomethine ylides and pyrrole-2,5-dione derivatives

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Abstract—Chiral C₂-symmetric diphenylthiophosphoramide ligand **L1** prepared from C₂-symmetric (1S,2S)-(-)-1,2-diphenylethylenediamine was found to be a fairly effective chiral ligand for Cu(I)-promoted 1,3-dipolar cycloaddition of imines and pyrrole-2,5-dione derivatives to give the corresponding adducts in moderate enantioselectivities and good yields. $© 2007 Elsevier Ltd. All rights reserved.$

1. Introduction

Previously we have reported that optically active diphenylphosphoramides, diphenylthiophosphoramides and diphenylselenophosphoramides of $(1R, 2R)$ - $(-)$ -1,2-diaminocyclohexane, $(1R,2R)$ -(+)-1,2-diphenylethylenediamine, or (R) -1,1'-binaphthyl-2,2'-diamine (BINAM) are easily available, relatively stable, recoverable, and effective chiral ligands in a variety of catalytic, asymmetric C–C bond formation reactions to give the corresponding products in good yields with high ee.^{[1](#page-5-0)} These results have promoted us to explore more catalytic, asymmetric reactions using these interesting chiral ligands. Herein we report that these chiral ligands are also fairly effective in the copper(I)-promoted 1,3-dipolar cycloaddition of azomethine ylides (from imines) and pyrrole-2,5-dione derivatives, which allows the stereoselective synthesis of pyrrolidines or proline derivatives in moderate enantio-selectivities.^{[2](#page-5-0)}

2. Results and discussion

Diphenylthiophosphoramides and diphenylselenophosphoramides L1–L4 as well as their derivatives L5–L6 were synthesized from the reaction of diphenylthiophosphinic or diphenylselenophosphinic chlorides as well as their analogues with $(1R, 2R)$ -(-)-1,2-diaminocyclohexane and $(1S, 2S)$ - $(-)$ -1,2-diphenylethylenediamine in the presence of diisopropylethylamine or triethylamine in dichloromethane, respectively $(Fig. 1)$ $(Fig. 1)$.^{[3](#page-5-0)} Initial examinations using imine 1a and 1-phenylpyrrole-2,5-dione 2a as the substrates in the presence of chiral diphenylthiophosphoramide ligand L1 and various metal salts were aimed at determining the optimal conditions, with the results of these experiments being summarized in [Table 1](#page-1-0). We found that using $Cu(CH_3CN)_4ClO_4$ (5 mol %) as a catalyst and Et₃N $(10 \text{ mol } \%)$ as a base gave the corresponding 1,3-dipolar cycloaddition product endo-3a, which was obtained in moderate to good yields and 5–55% ee in a variety of solvents in the presence of L1 (5.5 mol %) at 0° C [\(Table](#page-1-0) [1,](#page-1-0) entries $1-5$.^{[4](#page-5-0)} Dichloromethane is the solvent of choice. Next, we utilized DBU and ${}^{i}Pr_{2}NEt$ as bases for this reaction under otherwise identical conditions, and found that endo-3a was formed in 83% yield and 69% ee when using ⁱ ${}^{i}Pr_{2}NEt$ as the base [\(Table 1](#page-1-0), entries 6 and 7). Using either AgOAc or AgOTf (5 mol %) as a catalyst and Pr_2 NEt as

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Figure 1.

Table 1. Optimization of the reaction conditions in the 1,3-dipolar cycloaddition of azomethine ylides (from imines) and 1-phenylpyrrole-2,5-dione

	N Cl 1a	CO ₂ Me N ⁻ Ph 2a	Metal salts (5 mol%), Ligand (5.5 mol) Base (10 mol%), 0 °C, 24 h	СI	Pn. $^{\prime}$ CO ₂ Me $endo - 3a$	
Entry	Ligand	Lewis acid	Base	Solvent	Yield ^a $(\%)$	ee \rm^b (%)
	L1	$Cu(CH3CN)4ClO4$	Et ₃ N	Toluene	85	28
	L1	$Cu(CH_3CN)_4ClO_4$		THF	47	37
			Et ₃ N			
3	L1	$Cu(CH3CN)4ClO4$	Et ₃ N	CH ₃ CN	76	5
4	L1	$Cu(CH3CN)4ClO4$	Et ₃ N	Ether	83	9
5	L1	$Cu(CH3CN)4ClO4$	Et ₃ N	CH_2Cl_2	84	55
6	L1	$Cu(CH_3CN)_4ClO_4$	DBU	CH_2Cl_2	30	10
	L1	$Cu(CH3CN)4ClO4$	$P_{r2}NEt$	CH_2Cl_2	83	69
8	L1	AgOAc	Pr_2NEt	CH_2Cl_2	78	64
9	L1	AgOTf	r_2 NEt	CH_2Cl_2	54	69
10	L1	$Cu(OTf)_{2}$	$Pr_{2}NEt$	CH ₂ Cl ₂	Trace	
11	L1	$Zn(OTf)_2$	'Pr ₂ NEt	CH_2Cl_2	Trace	

^a Isolated yields.

^b Determined by chiral HPLC.

the base afforded endo-3a in 78% and 54% yields as well as 64% and 69% ee, respectively, in CH₂Cl₂ at 0 °C in the presence of L1 $(5.5 \text{ mol})\%$ (Table 1, entries 8 and 9). It should be noted that copper salts, $Cu(OTf)_2$ and $Zn(OTf)_2$, did not catalyze this reaction under the standard conditions (Table 1, entries 10 and 11). The best reaction conditions are to carry out the reaction in CH₂Cl₂ using Cu(CH₃CN)₄-ClO₄ (5 mol %) as a catalyst and $^{i}\text{Pr}_{2}$ NEt (10 mol %) as the base in the presence of L1 $(5.5 \text{ mol})\%$. In this reaction, P_{r_2} NEt acts as a base to generate the azomethine ylide.^{2e} The possible transition state of this reaction has been provided in [Figure 2](#page-2-0). We believe that diphenylthiophosphoramide ligand L1 is an (S,S)-bidentate ligand to the Cu center and its steric effect played a very important role for achieving higher ee.

With the optimized conditions in hand, we next examined the temperature effects on this reaction. As can be seen in [Table 2](#page-2-0), lowering the reaction temperature to -40° C

slightly improved the ee of 3a to 74%, although no improvement could be realized at -10 °C and -25 °C ([Ta](#page-2-0)[ble 2](#page-2-0), entries 1–3). Other chiral thiophosphoramide ligands L2, L4–L6 and chiral diphenylselenophosphoramide ligand L3 produced *endo*-3a in lower enantioselectivities under the standard conditions and no enantioselectivity could be observed if using L4 as a chiral ligand [\(Table 2,](#page-2-0) entries 4–8). Therefore, L1 is the best chiral ligand in this reaction.

 \mathbf{r}

The generality of this $Cu(CH_3CN)_4ClO_4/L1$ -catalyzed enantioselective 1,3-dipolar cycloaddition reaction was examined using a variety of imines and 1-methylpyrrole-2,5-dione, 1-benzylpyrrole-2,5-dione, and 1-phenylpyrrole-2,5-dione. The results are summarized in [Table 3.](#page-2-0) All reactions proceeded smoothly to give the corresponding products 3 in good yields and moderate ee under the optimal conditions [\(Table 3\)](#page-2-0). As for imine 1 bearing an electron-donating methyl group on the benzene ring or having no substituent on the benzene ring, the correspond-

Figure 2.

Table 2. Optimization of the reaction conditions in the 1,3-dipolar cycloaddition of azomethine ylides (from imines) and 1-phenylpyrrole-2,5-dione

^a Isolated yields.

^b Determined by chiral HPLC.

Table 3. Enantioselective 1,3-dipolar cycloaddition of azomethine ylides (from imines) and 1-phenylpyrrole-2,5-dione catalyzed by Cu(CH₃CN)₄ClO₄ in the presence of L1

	R ¹	CO ₂ R ² $\ddot{}$ Ő	Ligand $L1$ (5.5 mol%), $Cu(CH_3CN)_4ClO_4$ (5 mol%) $N-R^3$	ⁱ Pr ₂ NEt (10 mol%), CH ₂ Cl _{2,} -40 °C R ¹	$N_{\sim 0}$ O_{\leq} $^{\prime}CO_{2}R^{2}$ п	
		$\mathbf{2}$			endo-3	
Entry	R ¹	R^2	R ³	Product	Yield ^a $(\%)$	ee $^{\rm b}$ (%)
	p -Cl	Me	Ph	3a	82	74
	p -Br	Me	Ph	3 _b	85	58
	o -Cl	Me	Ph	3c	82	57
	H	Me	Ph	3d	89	47
	p -Me	Me	Ph	3e	76	26
	p -Cl	Et	Ph	3f	72	62
	$m-Br$	Me	Ph	3g	60	52
	H	Me	Me	3 _h	77	79
9	p -Cl	Me	Me	3i	86	77
10	p -Cl	Me	Benzyl	3j	83	68

^a Isolated yields.

^b Determined by chiral HPLC.

ing 1,3-dipolar cycloaddition adducts 3e and 3f were obtained in 47% and 26% ee, respectively (Table 3, entries 4 and 5). In other cases, 1,3-dipolar cycloaddition adducts 3 were formed in 55–77% ee (Table 3, entries 1–3, 6, and

 R^3

7). Using either 1-methylpyrrole-2,5-dione or 1-benzylpyrrole-2,5-dione as the substrate, the corresponding 1,3-dipolar cycloaddition adducts 3 were obtained in similar enantioselectivities ([Table 3](#page-2-0), entries 8–10).

3. Conclusion

In conclusion, chiral C_2 -symmetric diphenylthiophosphoramide L1, prepared from C_2 -symmetric $(1S, 2S)$ - $(-)$ -1,2diphenylethylenediamine, has been found to be a fairly effective chiral ligand for Cu(I)-promoted 1,3-dipolar cycloaddition of imines and pyrrole-2,5-dione derivatives to give the corresponding adducts in moderate enantioselectivities. These results will allow us to design and synthesize new effective sulfur-containing chiral ligands for this interesting asymmetric 1,3-dipolar cycloaddition reaction. Efforts are currently underway to elucidate the mechanistic details of this asymmetric 1,3-dipolar cycloaddition reaction and to disclose the exact structure of the active species in this catalytic system.

4. Experimental

4.1. General methods

Melting points were measured with a Yanagimoto micro melting point apparatus and are uncorrected. Optical rotations were determined in a solution of CHCl3 at 20 °C by using a Perkin–Elmer-241 MC polarimeter; [α]_D-values are given in units of 10^{-1} deg cm² g⁻¹. ¹H NMR spectra were recorded on a Bruker AM-300 spectrometer for solution in CDCl₃ with tetramethylsilane (TMS) as the internal standard; J-values are given in Hz. Mass spectra were recorded with a HP-5989 instrument and HRMS was measured by a Finnigan MA+ mass spectrometer. The organic solvents used were dried by standard methods when necessary. All solid compounds reported in this paper gave satisfactory CHN microanalyses with an Italian Carlo-Erba 1106 analyzer. Commercially obtained reagents were used without further purification. All reactions were monitored by TLC with Huanghai 60F₂₅₄ silica gel coated plates. Flash column chromatography was carried out using 300–400 mesh silica gel at increased pressure. All 1,3-dipolar cycloaddition reactions were performed under argon using standard Schlenk techniques. Enantiomeric excesses of sec-alcohols were determined by HPLC analysis using a chiral stationary phase column (column, Daicel Co. Chiralcel AD, AS, and OD) and the absolute configuration of the major enantiomer was assigned according to the sign of the specific rotation. Diphenylthiophosphinic and diphenylselenophosphinic chlorides were prepared upon heating chlorodiphenylphosphine with selenium at 120° C for 6 h. 1-Phenylpyrrole-2,5-dione 2a was synthesized according to the literature procedures.[5](#page-5-0)

4.2. General procedure for the synthesis of α -imino esters

To a suspension of the corresponding amino acid ester hydrochloride (23.9 mmol) and $MgSO₄$ (25.0 mmol) in CH_2Cl_2 (25 mL) was added Et₃N (3.4 mL, 23.9 mmol). The mixture was stirred at room temperature for 1 h, then the corresponding aldehyde (20.0 mmol) was added. The reaction was stirred at room temperature overnight, and then the resulting precipitate was removed by filtration. The filtrate was washed with water (15 mL), the aqueous phase was extracted with CH_2Cl_2 (10 mL) and the combined organic phases were washed with brine, dried over MgSO4, and concentrated under reduced pressure. The resulting pure imino esters were used in 1,3-dipolar cycloadditions without further purification.

4.3. General procedure for the preparation of chiral diphenylthiophosphoramides and diphenylselenophosphoramides $L1-L6$

To a solution of $(1S,2S)$ -(-)-1,2-diphenylethylenediamine (212 mg, 1.0 mmol) and triethylamine (303 mg, 3.0 mmol, 0.42 mL) in dichloromethane (20 mL) was added diphenylthiophosphinic chloride (504 mg, 2.0 mmol) at 0° C. After stirring the reaction mixture for 10 h, the solvent was removed under reduced pressure. The crude product was extracted with ether and washed with water $(3 \times 50 \text{ mL})$, 10% Na₂CO₃ (50 mL), and brine. The organic layer was dried over anhydrous $Na₂SO₄$ and then evaporated under reduced pressure. The residue was recrystallized from dichloromethane and hexane (4:1) to give L1 as colorless crystals (534 mg, 86%).

4.3.1. Ligand L1. This is a known compound.^{3a} Yield: $(534 \text{ mg}, 86\%)$. White solid. Mp: 230.0–230.3 °C. ¹H NMR (300 MHz, CDCl₃, TMS) δ 4.45–4.55 (m, 2H), 5.65–5.69 (m, 2H), 6.87 (d, $J = 7.8$ Hz, 4H), 7.03–7.15 (m, 10H), 7.26–7.29 (m, 2H), 7.40–7.46 (m, 4H), 7.50– 7.52 (m, 2H), 7.60–7.66 (m, 4H), 7.83–7.90 (m, 4H); ^{31}P NMR (CDCl₃, 121 MHz, 85% H₃PO₄) δ +66.0; $[\alpha]_D^{20} = +92.7$ (c 0.94, CH₂Cl₂).

4.3.2. Ligand L2. This is a known compound.^{3b} Yield: (475 mg, 89%). White solid. Mp: 139.9-140.0 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 1.05–1.60 (6H), 1.80– 1.87 (m, 2H), 3.26–3.36 (m, 2H), 4.00–4.04 (m, 2H), 7.31–7.37 (m, 4H), 7.44–7.51 (m, 8H), 7.76–7.83 (m, 4H), 8.02–8.09 (m, 4H); $[\alpha]_D^{20} = +20.8$ (c 1.28, CH₂Cl₂).

4.3.3. Ligand L3. This is a known compound.^{3c} Yield: (644 mg, 87%). White solid. Mp: 170-172 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 4.55–4.69 (m, 2H), 5.62–5.67 $(m, 2H)$, 6.89 (d, $J = 7.8$ Hz, 4H), 7.02–7.12 (m, 10H), 7.22–7.27 (m, 2H), 7.40–7.46 (m, 4H), 7.49–7.52 (m, 2H), 7.57–7.64 (m, 4H), 7.87–7.94 (m, 4H); $[\alpha]_D^{20} = +117.2$ (c $1.17, CH₂Cl₂$.

4.3.4. Ligand L4. Yield: (495 mg, 70%). White solid. Mp: 130.0–130.1 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 4.27–4.34 (m, 2H), 4.90–4.98 (m, 2H), 6.89–6.92 (m, 4H), 6.95–6.98 (m, 4H), 7.05–7.29 (m, 22H); 13C NMR (CDCl3, TMS, 75 MHz) δ 61.6 (d, $J_{\text{C-P}} = 7.5$ Hz), 121.03, 121.09, 121.15, 121.2, 125.06, 125.08, 125.22, 125.24, 127.98, 128.02, 128.3, 129.32, 129.34, 129.49, 129.51, 138.5 (d, $J_{\text{C-P}} = 3.5$ Hz), 150.6 (d, $J_{\text{C-P}} = 7.5$ Hz), 150.8 (d, $J_{\text{C-P}} =$ 8.0 Hz); ³¹P NMR (CDCl₃, 121 MHz, 85% H₃PO₄) δ

+63.5; IR (CHCl₃) v 1590, 1489, 1455, 1188, 918 cm⁻¹; HRMS(ESI) calcd for $C_{38}H_{35}N_2O_4P_2S_2$ (M+H⁺): 709.1514, found: 709.1513. $[\alpha]_{\text{D}}^{20} = -47.8 \ (\text{c} \ 0.86, \text{CH}_2\text{Cl}_2).$

4.3.5. Ligand L5. Yield: (412 mg, 80%). White solid. Mp: 70.2–70.9 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 1.02 (t, $J = 6.9$ Hz, 6H), 1.26 (t, $J = 6.9$ Hz, 6H), 3.53–3.64 (m, 2H), 3.86–4.02 (m, 8H), 4.44–4.55 (m, 2H), 6.95–6.98 (m, 4H), 7.18–7.28 (m, 6H); ¹³C NMR (CDCl₃, TMS, 75 MHz) δ 15.4 (d, $J_{C-P} = 9.2$ Hz), 15.8 (d, $J_{C-P} =$ 8.6 Hz), 61.1 (d, $J_{\text{C-P}} = 6.8$ Hz), 62.9 (d, $J_{\text{C-P}} = 4.7$ Hz), 63.1 (d, $J_{\text{C-P}} = 5.2 \text{ Hz}$), 127.51, 127.55, 128.0, 139.3 (d, $J_{\text{C-P}} = 2.7 \text{ Hz}$); ³¹P NMR (CDCl₃, 121 MHz, 85% H_3PO_4) δ +71.6; IR (CH₂Cl₂) v 2970, 2839, 1738, 1454, 1376, 1167, 972 cm⁻¹; HRMS(ESI) calcd for $C_{22}H_{35}$ - $N_2O_4P_2S_2$ (M+H⁺): 517.1514, found: 517.1515. $[\alpha]_D^{20} = -18.8$ (c 1.16, CH₂Cl₂).

4.3.6. Ligand L6. Yield: (352 mg, 78%). White solid. Mp: 154.7–155.8 °C; ¹H NMR (CDCI₃, TMS, 300 MHz) δ 0.76 (dt, $J = 7.5$, 20.1 Hz, 6H), 1.31 (dt, $J = 7.5$, 19.5 Hz, 6H), 1.59–1.71 (m, 4H), 2.09–2.21 (m, 4H), 4.41–4.49 (m, 2H), 4.76–4.80 (m, 2H), 6.93–6.97 (m, 4H), 7.11–7.13 (m, 6H); ¹³C NMR (CDCl₃, TMS, 75 MHz) δ 6.8 (d, $J_{C-P} =$ 3.2 Hz), 6.9 (d, $J_{\text{C-P}} = 4.4$ Hz), 24.0 (d, $J_{\text{C-P}} = 6-5.1$ Hz), 27.2 (d, $J_{\text{C-P}} = 68.4 \text{ Hz}$), 60.7 (d, $J_{\text{C-P}} = 5.1 \text{ Hz}$), 127.1, 127.7, 127.9, 141.3 (d, $J_{C-P} = 3.8$ Hz); ³¹P NMR (CDCl₃, 121 MHz, 85% H₃PO₄) δ +82.0; IR (CHCl₃) v 2970, 2841, 1738, 1455, 1376, 1167, 973 cm-1 ; HRMS(ESI) calcd for $C_{22}H_{35}N_2P_2S_2$ (M+H⁺): 453.1717, found: 453.1714. $[\alpha]_D^{20} = +76.5$ (c 0.87, CH₂Cl₂).

4.4. Typical reaction procedure

The catalyst was prepared by stirring $Cu(CH_3CN)_4ClO_4$ (4.9 mg, 0.015 mmol, 5 mol %) and ligand L1 (11.9 mg, 0.0165 mmol, 5.5 mol %) in CH_2Cl_2 (2 mL) for 1 h at room temperature. The reaction mixture was then cooled to -40 °C, after which imine substrate 1 (0.30 mmol), 1-phenylpyrrole-2,5-dione 2 (0.45 mmol, 1.5 equiv) and base $(0.03 \text{ mmol}, 10 \text{ mol})$ % were subsequently added and the resulting mixture was stirred at $-40\degree$ C for 24 h. When the reaction was complete, as monitored by TLC, the corresponding pure adduct was purified by column chromatography on silica gel (petroleum ether/ethyl acetate/1% triethylamine).

4.5. (1S,3R,3aS,6aR)-Methyl-4,6-dioxo-3-(p-chlorophenyl)- 5-phenyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3a

This is a known compound.⁶ A white solid. Mp: 140.7– 148.9 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 2.48–2.50 (m, 1H), 3.56 (t, $J = 8.1$ Hz, 1H), 3.74 (t, $J = 7.2$ Hz, 1H), 3.87 (s, 3H), 4.14 (dd, $J = 4.8$, 6.6 Hz, 1H), 4.58 (dd, $J = 4.8$, 7.8 Hz, 1H), 7.14 (d, $J = 7.8$ Hz, 2H), 7.31– 7.43 (m, 7H); $[\alpha]_D^{20} = +91.5$ (c 0.80, CH₂Cl₂) for 74% ee; Chiralcel AS-H, hexane/ i PrOH = 50:50, 1.5 mL/min, 220 nm, $t_{\text{major}} = 8.33 \text{ min}, t_{\text{minor}} = 30.61 \text{ min}.$

4.6. (1S,3R,3aS,6aR)-Methyl-4,6-dioxo-3-(p-bromophenyl)- 5-phenyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3b

A white solid. Mp: 175.6–176.7 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 2.47–2.50 (m, 1H), 3.56 (t, J = 8.4 Hz, 1H), 3.75 (t, $J = 7.2$ Hz, 1H), 3.87 (s, 3H), 4.11–4.16 (m, 1H), 4.56 (dd, $J = 4.8$, 8.4 Hz, 1H), 7.12–7.15 (m, 2H), 7.31–7.50 (m, 7H); ¹³C NMR (CDCl₃, TMS, 75 MHz) δ 47.9, 49.0, 52.3, 61.7, 63.3, 122.2, 126.0, 128.6, 128.8, 129.1, 131.4, 131.5, 135.8, 169.9, 173.5, 174.9; IR (CH_2Cl_2) m 3333, 2948, 2837, 1747, 1713, 1489, 1381, 1206, 1010 cm⁻¹; MS (ESI) m/z : 451 [M+Na⁺]. Anal. Calcd for $C_{20}H_{17}BrN_2O_4$ requires: C, 55.96; H, 3.99; N, 6.53. Found: C, 56.66; H, 3.64; N, 6.25. $[\alpha]_D^{20} = +76.4$ (c 1.07, CH₂Cl₂), for 58% ee; Chiralcel AD-H, hexane/ $iPrOH = 50:50$, 0.7 mL/min, 230 nm, $t_{\text{minor}} = 12.14 \text{ min}$, $t_{\text{major}} = 24.24 \text{ min}$.

4.7. (1S,3R,3aS,6aR)-Methyl-4,6-dioxo-3-(o-chlorophenyl)- 5-phenyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3c

A white solid. Mp: 153.0–154.8 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 2.41–2.44 (m, 1H), 3.75 (t, $J = 7.2$ Hz, 1H), 3.84–3.90 (m, 4H), 4.18 (dd, $J = 4.8$, 6.3 Hz, 1H), 4.88 (dd, $J = 4.8$, 8.1 Hz, 1H), 7.08 (d, $J = 8.4$ Hz, 2H), 7.25–7.43 (m, 6H), 7.67–7.70 (m, 1H); ¹³C NMR (CDCl₃, TMS, 75 MHz) d 46.5, 47.7, 52.3, 60.5, 61.4, 126.0, 126.9, 127.2, 128.4, 128.9, 129.2, 131.5, 133.3, 134.9, 144.9, 170.0, 173.3, 175.1; IR $\text{(CH}_2\text{Cl}_2)$ v 3333, 1748, 1713, 1500, 1383, 1206 cm-1 ; HRMS (ESI) calcd for $C_{20}H_{17}CIN_2O_4$ (M+Na⁺): 407.0775, found: 407.0770. $[\alpha]_D^{20} = +42.2$ (c 0.83, CH₂Cl₂) for 57% ee; Chiralcel AD-H, hexane/iPrOH = 50:50, 0.7 mL/min, 230 nm, t_{major} = 11.93 min, $t_{\text{minor}} = 30.13 \text{ min}$.

4.8. (1S,3R,3aS,6aR)-Methyl-4,6-dioxo-3,5-diphenyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3d

This is a known compound.^{[4](#page-5-0)} A white solid. Mp: 178.2– 180.5 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 2.51–2.54 (m, 1H), 3.57 (t, $J = 8.4$ Hz, 1H), 3.73 (t, $J = 7.2$ Hz, 1H), 3.87 (s, 3H), 4.12–4.16 (m, 1H), 4.61 (dd, $J = 5.1$, 8.4 Hz, 1H), 7.14 (d, $J = 7.2$ Hz, 2H), 7.29–7.47 (m, 8H); $[\alpha]_D^{20} = +44.1$ (c 0.73, CH₂Cl₂) for 47% ee; Chiralcel AD-H, hexane/*i*PrOH = 50:50, 0.7 mL/min, 230 nm, $t_{\text{minor}} =$ 13.79 min, $t_{\text{major}} = 22.21$ min.

4.9. (1S,3R,3aS,6aR)-Methyl-4,6-dioxo-3-(p-methylphenyl)- 5-phenyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3e

A white solid. Mp: 176.9–177.5 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 2.33 (s, 3H), 2.49 (br, 1H), 3.54 (t, $J = 8.4$ Hz, 1H), 3.72 (t, $J = 7.2$ Hz, 1H), 3.87 (s, 3H), 4.11–4.15 (m, 1H), 4.58 (dd, $J = 4.8$, 8.7 Hz, 1H), 7.14– 7.18 (m, $\overline{4H}$), 7.29–7.42 (m, 5H); ¹³C NMR (CDCl₃, TMS, 75 MHz) d 21.2, 48.3, 49.3, 52.3, 61.8, 64.1, 126.1, 126.9, 128.4, 129.0, 129.2, 131.5, 133.5, 138.0, 170.1, 173.7, 175.2; IR (CH₂Cl₂) v 3333, 2956, 2844, 1748, 1714, 1500, 1382, 1206 cm⁻¹; MS (ESI) m/z : 387 [M+Na⁺]. Anal. Calcd for $C_{21}H_{20}N_2O_4$ requires: C, 69.22; H, 5.53, N,

7.28%. Found: C, 69.13; H, 5.29, N, 7.49. $\left[\alpha\right]_{D}^{20} = +42.2$ (c) 0.83, CH_2Cl_2), for 26% ee; Chiralcel AD-H, hexane/ $iPfOH = 50:50$, 0.7 mL/min, 230 nm, $t_{\text{minor}} = 14.09$ min, $t_{\text{major}} = 30.21 \text{ min.}$

4.10. (1S,3R,3aS,6aR)-Ethyl-4,6-dioxo-3-(p-chlorophenyl)- 5-phenyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3f

A white solid. Mp: 75.7–76.9 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 1.37 (t, $J = 7.2$ Hz, 1H), 2.48–2.50 (m, 1H), 3.56 (t, $J = 8.1$ Hz, 1H), 3.74 (t, $J = 7.5$ Hz, 1H), 4.12 (dd, $J = 4.8$, 6.3 Hz, 1H), 4.30–4.38 (m, 2H), 4.57 (dd, $J = 4.8, 6.0$ Hz, 1H), 7.14 (d, $J = 8.1$ Hz, 2H), 7.31–7.42 (m, 7H); ¹³C NMR (CDCl₃, TMS, 75 MHz) δ 14.1, 47.9, 49.2, 61.5, 61.9, 63.4, 126.0, 128.4, 128.56, 128.63, 129.1, 131.4, 134.0, 135.2, 169.4, 169.8, 173.5, 174.8; IR (CH_2Cl_2) v 3336, 2983, 2844, 1716, 1597, 1492, 1381, 1205 cm⁻¹; MS (ESI) m/z : 421 [M+Na⁺]. Anal. Calcd for C₂₁H₁₉ClN₂O₄ requires: C, 63.24; H, 4.80; N, 7.02. Found: C, 63.42; H, 4.55; N, 6.94. $[\alpha]_D^{20} = +111.5$ (c 1.08, CH₂Cl₂) for 62% ee; Chiralcel OD-H, hexane/iPrOH = 50:50, 0.7 mL/min, 230 nm, $t_{\text{major}} = 18.93 \text{ min}, t_{\text{minor}} = 26.58 \text{ min}.$

4.11. (1S,3R,3aS,6aR)-Methyl-4,6-dioxo-3-(m-bromophenyl)-5-phenyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3g

A white solid. Mp: 157.2–157.3 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 2.51 (br, 1H), 3.56 (t, $J = 8.4$ Hz, 1H), 3.74 (t, $J = 6.9$ Hz, 1H) 3.87 (s, 3H), 4.13 (dd, $J = 4.5$, 6.6 Hz, 1H), 4.57 (dd, $J = 4.5$, 9.0 Hz, 1H), 7.14–7.24 (m, 3H), 7.34–7.46 (m, 5H), 7.67 (s, 1H); ¹³C NMR (CDCl₃, TMS, 75 MHz) d 47.9, 49.0, 52.3, 61.6, 63.2, 122.7, 126.0, 126.1, 128.6, 129.1, 130.0, 131.4, 131.5, 139.3, 145.3, 169.8, 173.5, 174.9; IR (CH₂Cl₂) v 3337, 2948, 2837, 1747, 1713, 1500, 1382, 1206 cm⁻¹; MS (ESI) m/z : 451 [M+Na⁺]. Anal. Calcd for $C_{20}H_{17}BrN_2O_4$ requires: C, 55.96; H, 3.99; N, 6.53. Found: C, 56.05; H, 3.67; N, 6.36. $[\alpha]_D^{20} = +58.3$ (c 0.80, CH₂Cl₂) for 52% ee; Chiralcel OD-H, hexane/iPrOH = 50:50, 1 mL/min, 230 nm, t_{major} = 19.51 min, $t_{\text{minor}} = 36.26 \text{ min.}$

4.12. (1S,3R,3aS,6aR)-Methyl-5-methyl-4,6-dioxo-3-phenyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3h

This is a known compound.⁴ A white solid. Mp: $152.1-$ 153.7 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 2.42 (br, 1H), 2.88 (s, 3H), 3.43 (t, $J = 8.4$ Hz, 1H), 3.57 (t, $J = 7.2$ Hz, 1H), 3.89 (s, 3H), 4.04–4.08 (m, 1H), 4.50 (dd, $J = 4.8$, 8.7 Hz, 1H), 7.35 (s, 5H); $[\alpha]_D^{20} = +30.5$ (c 0.60, CH_2Cl_2) for 79% ee; Chiralcel AD-H, hexane/ $iPrOH = 50:50, 0.7 \text{ mL/min}, 220 \text{ nm}, t_{\text{major}} = 29.54 \text{ min},$ $t_{\rm minor} = 33.29$ min.

4.13. (1S,3R,3aS,6aR)-Methyl-5-methyl-4,6-dioxo-3-(pchlorophenyl)-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3i

A white solid. Mp: 210.7–211.1 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) $\bar{\delta}$ 2.32 (br, 1H), 2.81 (s, 3H), 3.35 (t, $J = 8.1$ Hz, 1H), 3.50 (t, $J = 7.2$ Hz, 1H), 3.81 (s, 3H), 3.98 (d, $J = 6.6$ Hz, 1H), 4.40 (d, $J = 8.7$ Hz, 1H), 7.20– 7.27 (m, 4H); ¹³C NMR (CDCl₃, TMS, 75 MHz) δ 25.0, 47.8, 49.1, 52.3, 61.4, 63.1, 128.4, 128.6, 133.9, 135.2, 170.0, 174.5, 175.8; IR (CH₂Cl₂) v 3341, 2933, 1749, 1698, 1437, 1215 cm⁻¹; MS (ESI) m/z : 345 [M+Na⁺]. Anal. Calcd for $C_{15}H_{15}C1N_2O_4$ requires: C, 55.82; H, 4.68; N, 8.68. Found: C, 55.73; H, 4.53; N, 8.54. $[\alpha]_D^{20} = +86.2$ (c 0.75, CH_2Cl_2) for 77% ee; Chiralcel AS-H, hexane/ $iPrOH = 50:50$, 1.5 mL/min, 220 nm, $t_{\text{major}} = 7.40$ min, $t_{\text{minor}} = 35.47 \text{ min.}$

4.14. (1S,3R,3aS,6aR)-Methyl-4,6-dioxo-3-(p-chlorophenyl)-5-benzyl-octahydropyrrole[3,4-c]pyrrole-1-carboxylate endo-3j

A white solid. Mp: 155.4–155.7 °C; ¹H NMR (CDCl₃, TMS, 300 MHz) δ 2.29 (br, 1H), 3.37 (t, $J = 8.1$ Hz, 1H), 3.56 (t, $J = 7.2$ Hz, 1H), 3.88 (s, 3H), 4.04 (d, $J = 7.2$ Hz, 1H), 4.41–4.61 (m, 3H), 7.04–7.17 (m, 4H), 7.31 (s, 5H); ¹³C NMR (CDCl₃, TMS, 75 MHz) δ 42.6, 47.9, 49.0, 52.3, 61.6, 63.3, 127.9, 128.42, 128.44, 128.5, 129.0, 133.7, 134.9, 135.5, 169.9, 174.1, 175.4; IR (CH_2Cl_2) v 3348, 2956, 5852, 1749, 1704, 1398, 1207 cm⁻¹; MS (ESI) m/z . 421 [M+Na⁺]. Anal. Calcd for $C_{21}H_{19}C1N_2O_4$ requires: C, 63.24; H, 4.80; N, 7.02. Found: C, 63.07, H, 4.68, N, 6.83. $[\alpha]_D^{20} = +69.5$ (c 0.62, CH₂Cl₂), for 68% ee; Chiralcel AS-H, hexane/*i*PrOH = 50:50, 1.5 mL/min, 220 nm, $t_{\text{major}} = 7.04 \text{ min}, t_{\text{minor}} = 25.33 \text{ min}.$

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References

- 1. Zhang, W.; Shi, M. Synlett 2007, 19–30, and references cited therein.
- 2. For recent reviews of 1,3-dipolar cycloaddition reactions of azomethine ylides, see: (a) Nájera, C.; Sansano, J. M. Curr. Org. Chem. 2003, 7, 1105–1150; (b) Synthetic Applications of 1,3-Dipolar Cycloaddition Chemistry Towards Heterocycles and Natural Products; Padwa, A., Pearson, W. H., Eds.; Wiley: New York, 2003; (c) Kanemasa, S. Synlett 2002, 1371–1387; (d) Gothelf, K. V. In Cycloaddition Reactions in Organic Synthesis; Kobayashi, S., Jørgensen, K. A., Eds.; Wiley-VCH: Weinheim, 2002; pp 211-245; (e) Nájera, C.; Sansano, J. M. Angew. Chem., Int. Ed. 2005, 44, 6272–6276.
- 3. (a) Shi, M.; Sui, W.-S. Tetrahedron: Asymmetry 1999, 10, 3319–3325; (b) Shi, M.; Wu, X.-F.; Rong, G. Chirality 2002, 14, 90–95; (c) Shi, M.; Sui, W.-S. Tetrahedron: Asymmetry 2000, 11, 835–841.
- 4. The endo-configuration of $3a$ was determined by ¹H NMR spectroscopic data reported in Cabrera, S.; Arrayás, R. G.; Carretero, J. C. J. Am. Chem. Soc. 2005, 127, 16394-16395.
- 5. Yoshitake, Y.; Misaka, J.; Setoguchi, K.; Abe, M.; Kawaji, T.; Eto, M.; Harano, K. J. Chem. Soc., Perkin Trans. 2 2002, 1611–1619.
- 6. Oderaotoshi, Y.; Cheng, W.-J.; Fujitomi, S.; Kasano, Y.; Minakata, S.; Komatsu, M. Org. Lett. 2003, 5, 5043–5046.